

High-density confinement of H₂ in ultramicroporous carbon materials



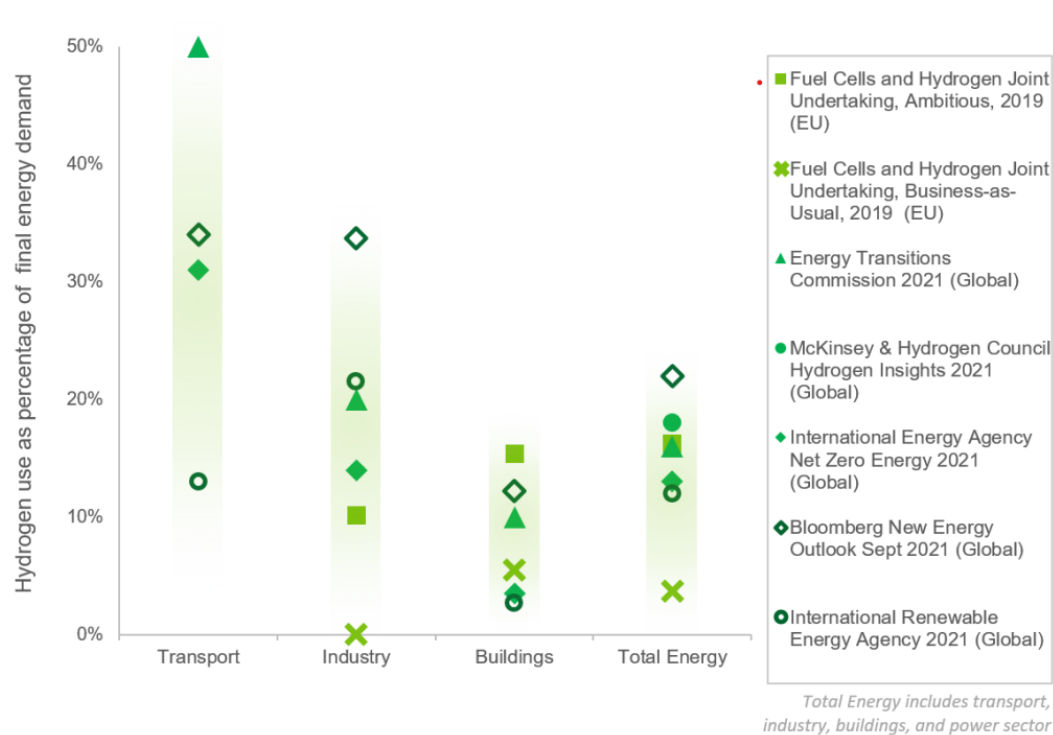
Assoc. Prof. Rasmus Palm

Institute of Chemistry
University of Tartu
Tartu, Estonia
rasmus.palm@ut.ee

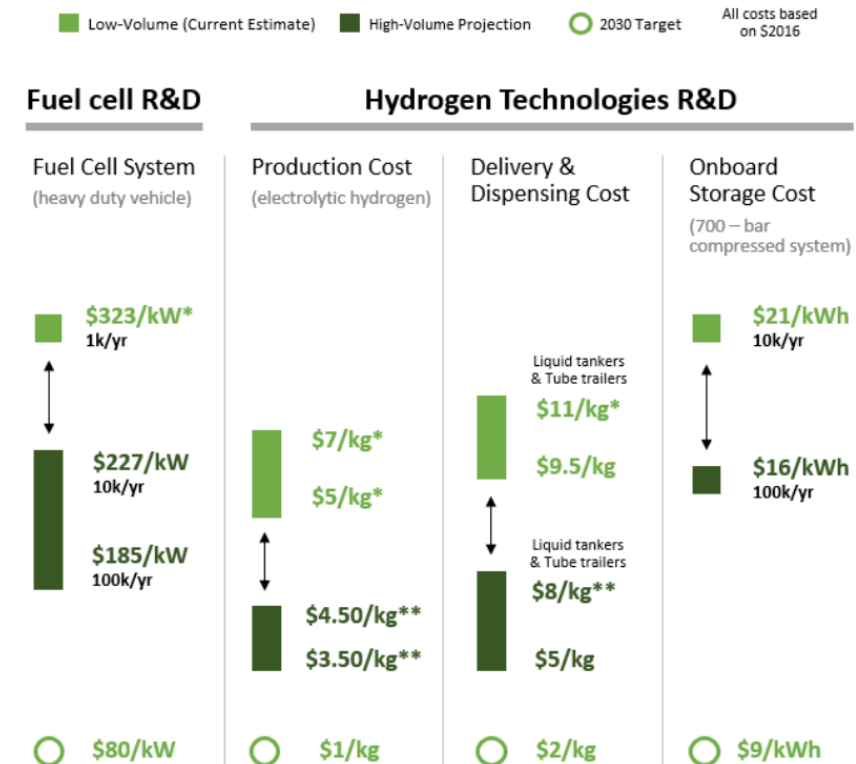


H₂ and its future role in economy

- H₂-based or -supported systems vital for greenhouse gas emission free economy
- A global need of at least 100 Mt of H₂ production by 2030 envisioned to meet climate goals
- 10 Mt + 10 Mt (H₂ production + import) in EU aimed by 2030 through REPowerEU Plan By 2050, renewable H₂ to cover around 10% of EU's energy needs.



U.S. National Clean Hydrogen Strategy and Roadmap, 2023



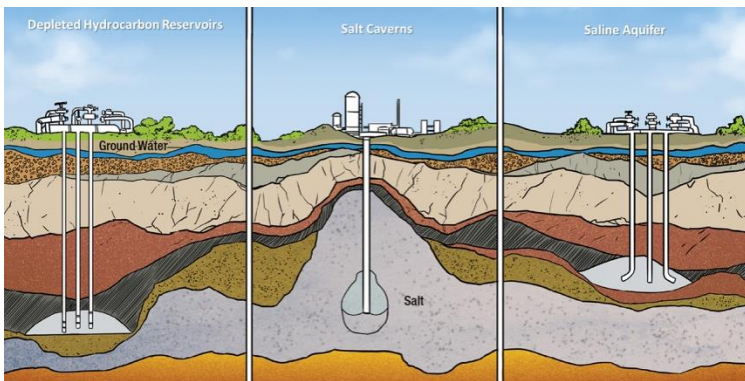
H₂ storage – staying stationary or going mobile

- Compressed (41 g_{H2}/L at 273.15 K and 700 bar)
Used for mobile applications
Compression and tank material requirements
- Liquefied (70 g_{H2}/L at 20 K)
Mobile and stationary applications
Liquefaction losses, boiloff
- Underground storage (up to 15 g_{H2}/L at 350 bar)
Stationary applications
Geographically restricted

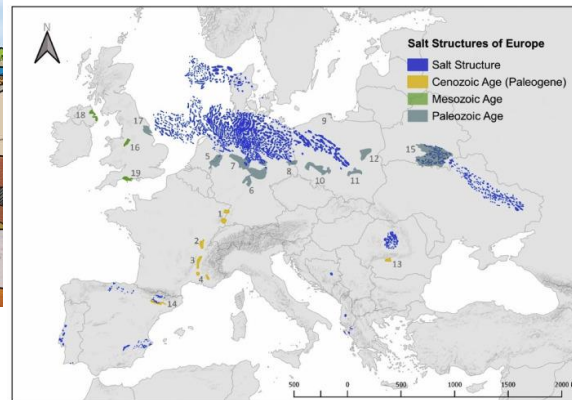


Type IV H₂ tank for 700 bar (left) and 60 bar (right) storage

M. Fiekiewicz et al., Hydrogen Onboard Storage, HYTEC, QUADRHY technical datasheet Technologies for Vehicles, 2023



B. N. Tackie-Otoo et al., Fuel 356(15), 2024, 129609



D. G. Caglayan et al., Int. J. Hydrogen Energy 45(11), 2020, 6793



Liquid H₂ tanker Suiso Frontier



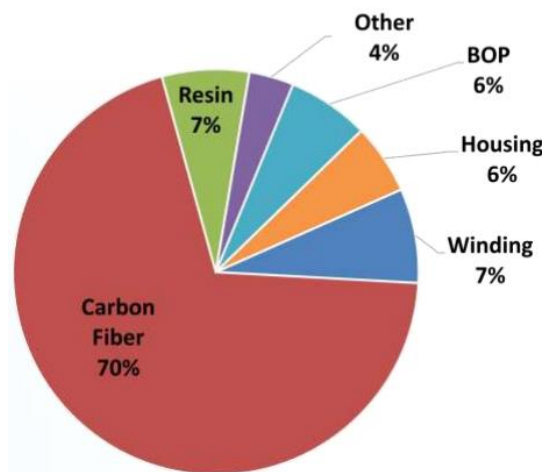
Liquefied H₂ tank at Kennedy NASA space center

<https://www.energy.gov/sites/default/files/2021-10/doe-perspectives-lh2.pdf>

H₂ storage – cost of using contemporary solutions

Current H₂ storage situation:

- „Cheap“ solutions are localized
- Mobile solutions with high H₂ density are expensive

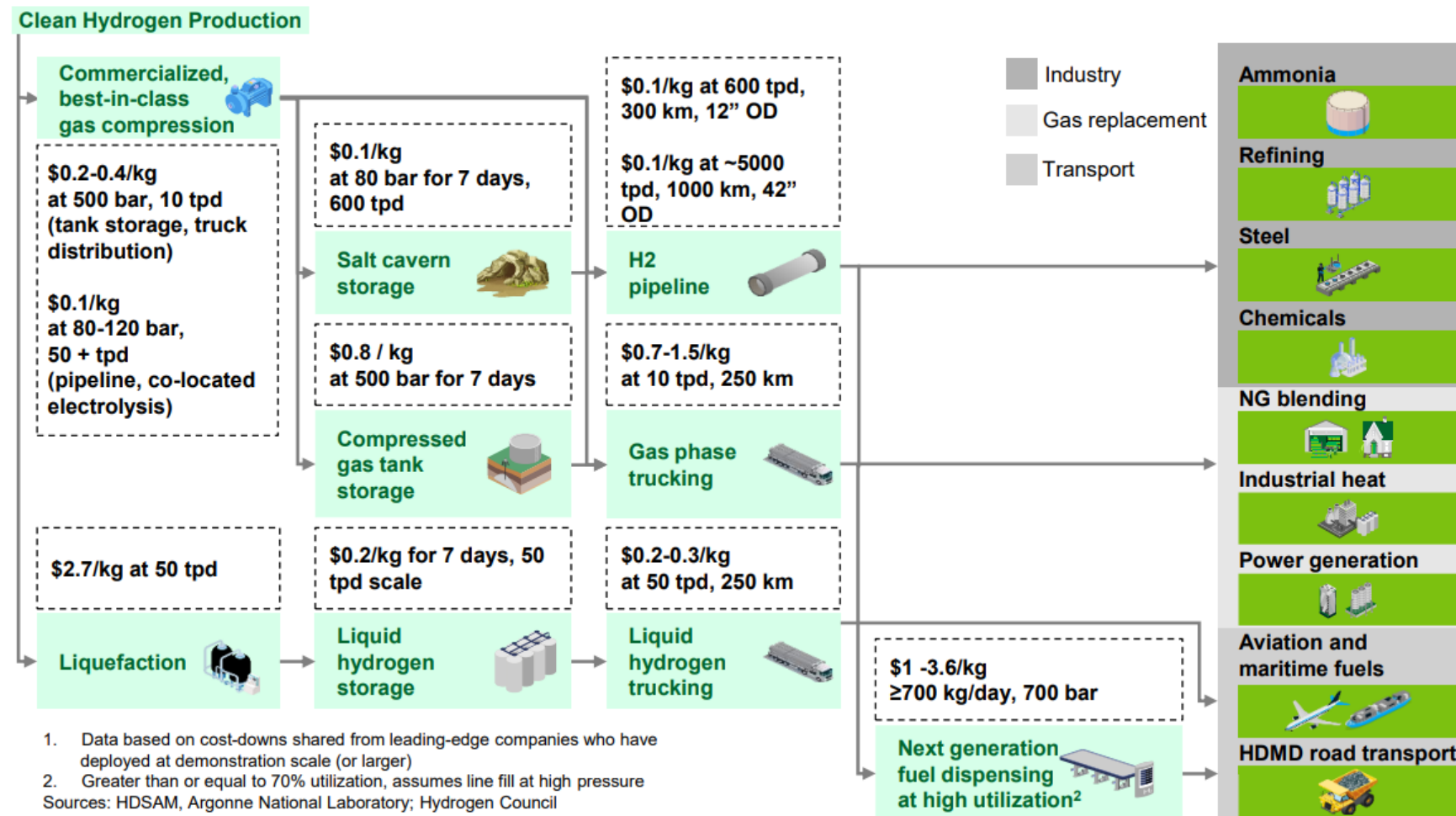


Cost breakdown of a Type 4 700 bar H₂ tank

C. Houchins et al., Hydrogen Storage Cost Analysis, 2022

2030 Midstream costs if advances in distribution and storage technology are commercialized:
Hydrogen distribution and storage assuming state-of-art technology at scale¹

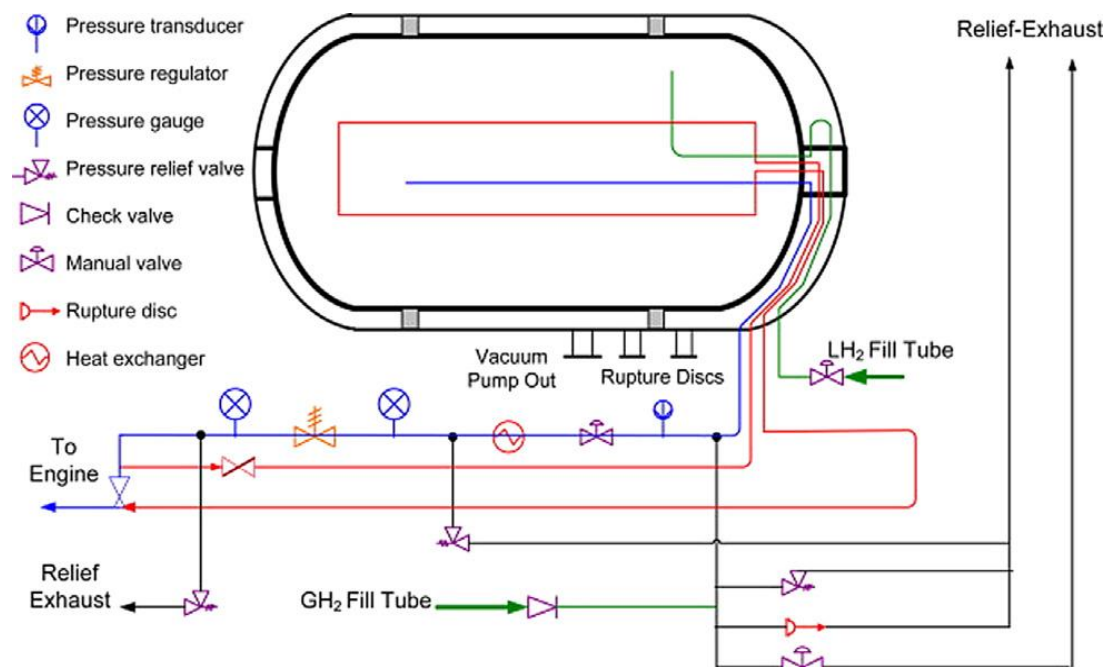
Downstream: E



U.S. National Clean Hydrogen Strategy and Roadmap, 2023

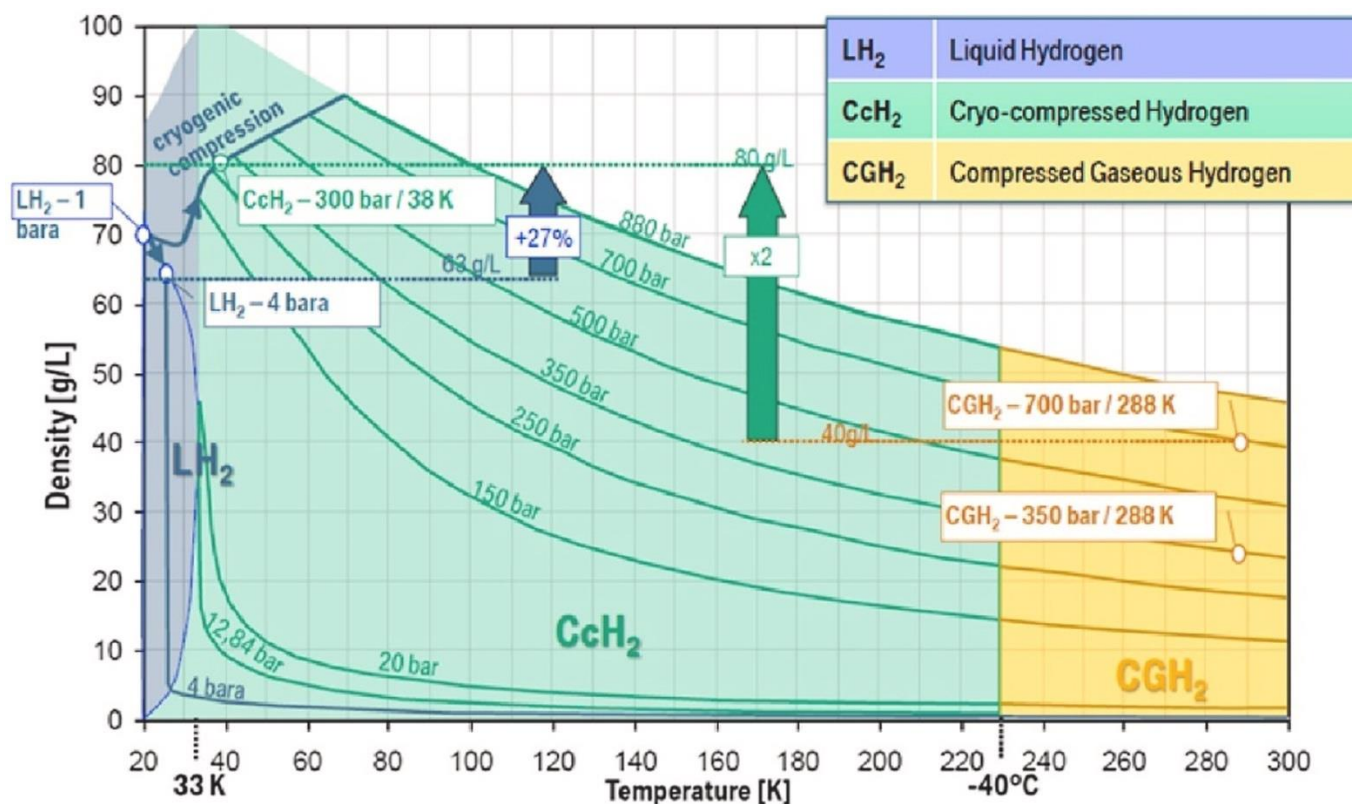
Cryocompression of H₂

- Combination of compression and cooled-down H₂
- Up to 80 g_{H2} L⁻¹
- Needs additional cooling over time or venting



Design schematic of a cryo-compressed H₂ storage system

G. Petitpas et al., Int. J. Hydrogen Energy 39, **2014**, 10564

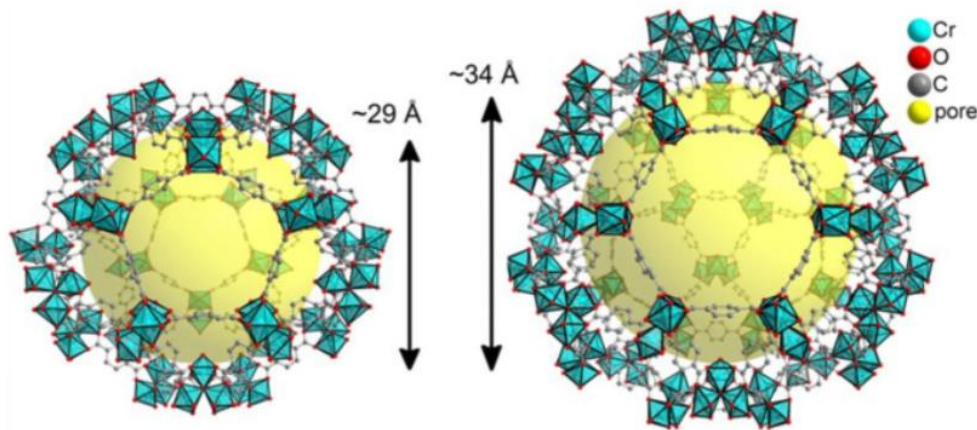


H₂ density vs pressure and temperature and regions corresponding to storage methods with pressure limits claimed by BMW in 2012

T. Zhang et al., Renewable Sustainable Energy Rev. 176, **2023**, 113204

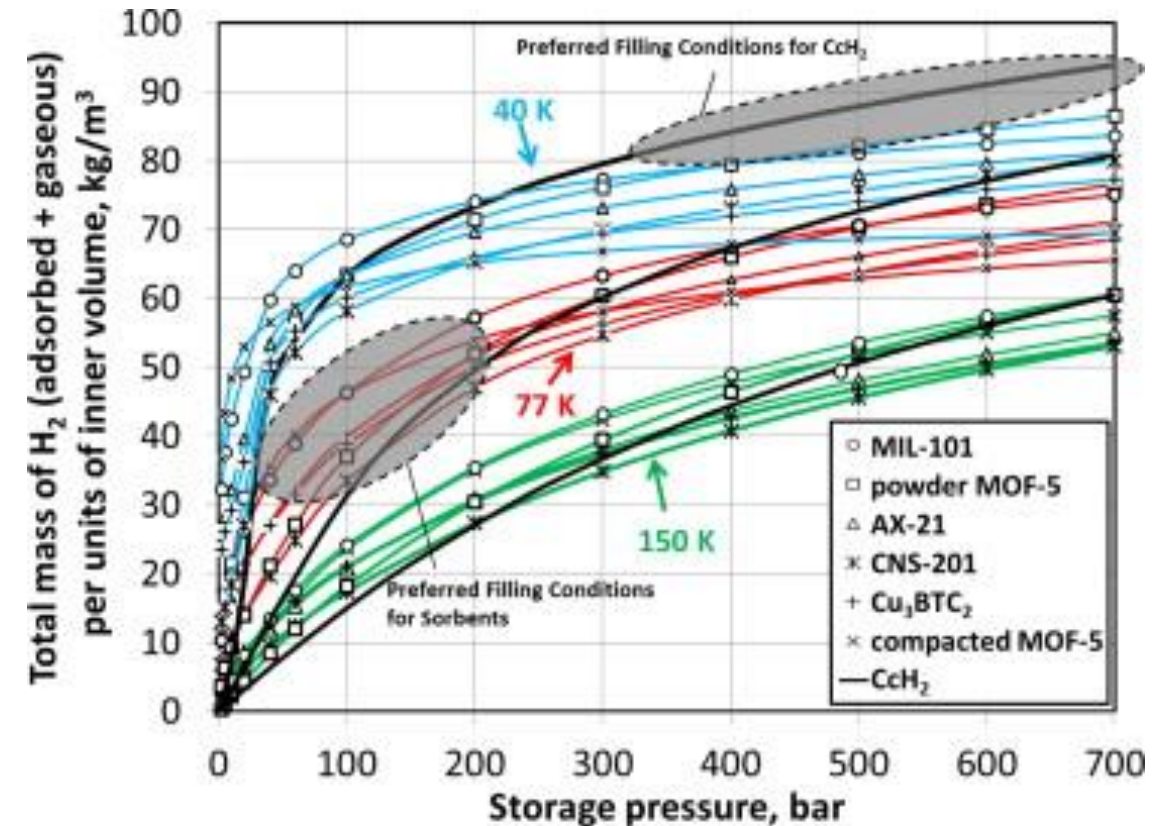
Cryoadsorption of H₂

- Enhanced H₂ density from adsorption inside pores of a solid material
- Effect scales with temperature and pressure, like for compression
- Amount and characteristic of pores of critical importance for H₂ density



Two variety of crystalline structures of MIL-101

M. Zou et al., Int. J. Mol. Sci. 23(16), **2022**, 9396

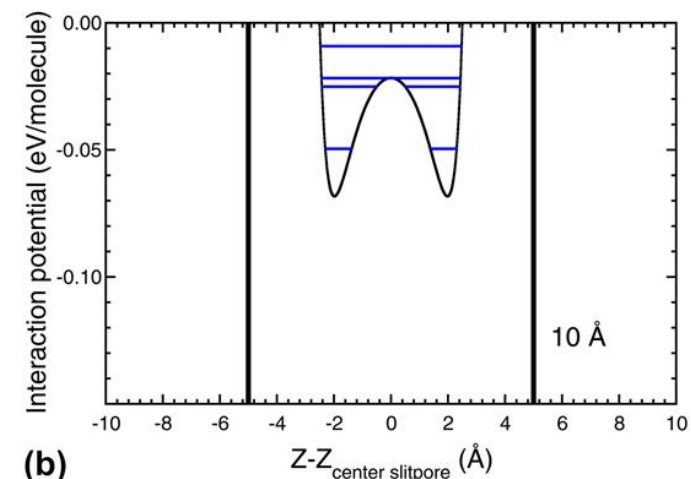
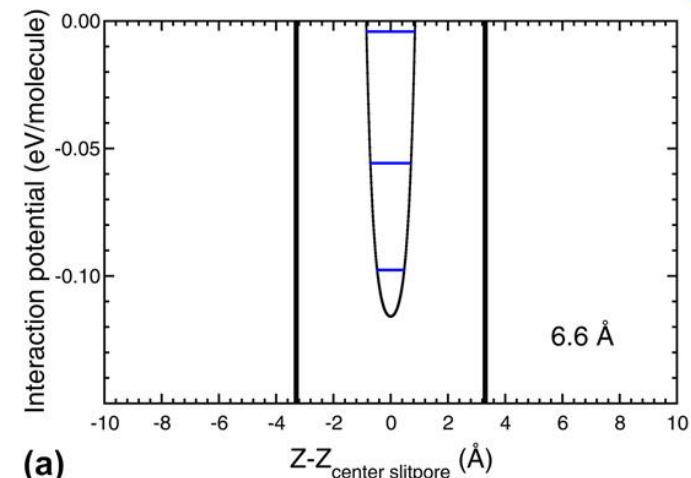
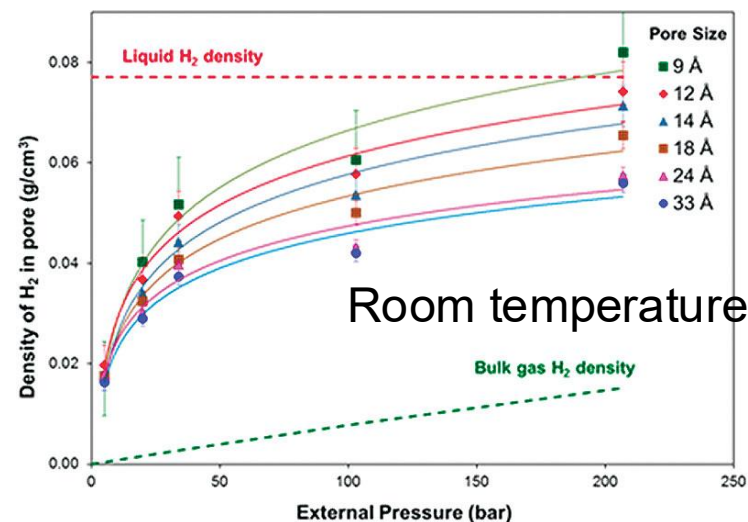
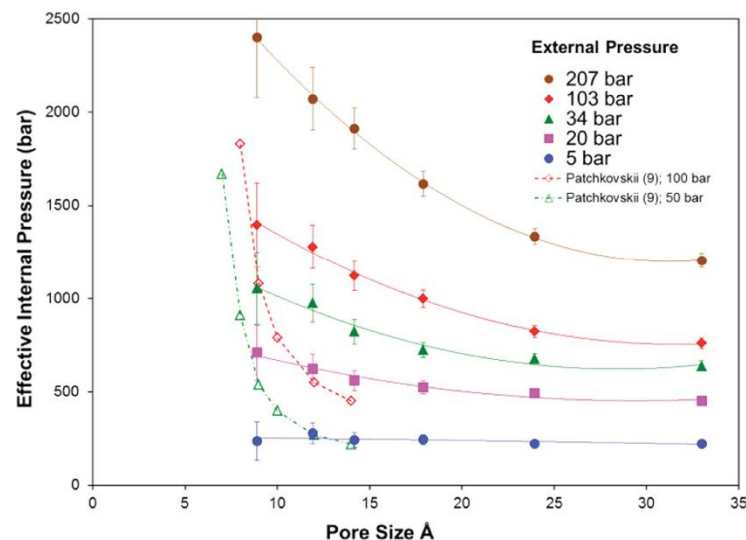


Volumetric H₂ density at different T and p conditions for cryocompression (CcH₂) and –adsorption systems

G. Petitpas et al., Int. J. Hydrogen Energy 39(20), **2014**, 10564

Cryoadsorption of H₂

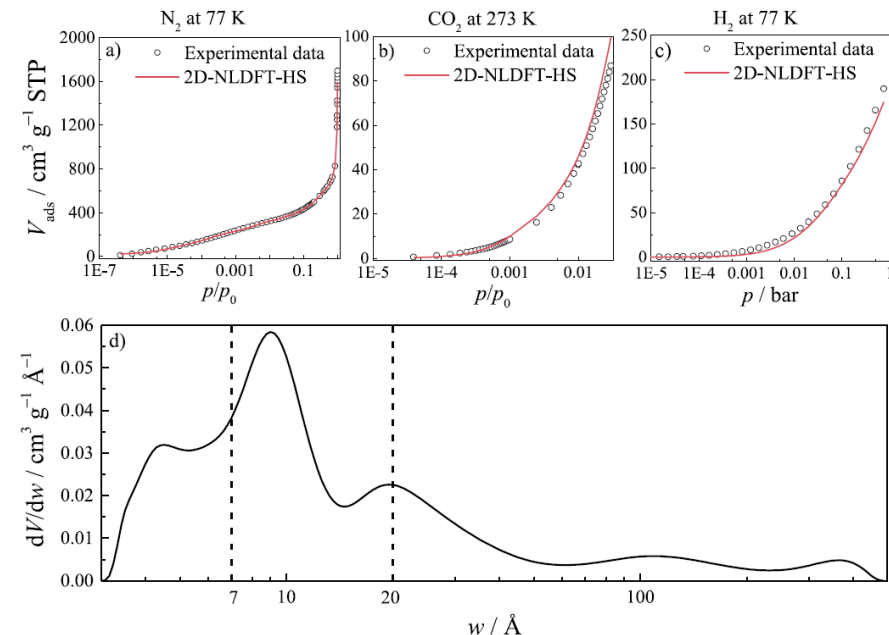
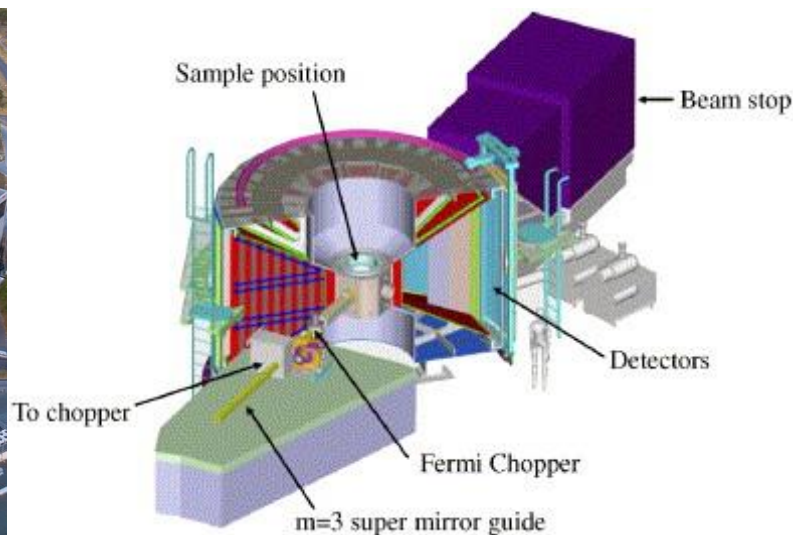
- Pores with optimal size:
 - Confine H₂ effectively
 - Yield high H₂ densities at increased temperatures
- Densities beyond that of liquid H₂ have been shown at room temperature
- Current limitations:
 - Nonscalable and noneconomic adsorbents
 - Nonuniform optimal porosity throughout adsorbent



H₂ interaction potential modelled for pores with different widths

Looking for the solution

- Fundamental investigation of H_2 dynamics in different porous structure
 - Labsale and neutron scattering methods for full characterisation



ISIS Neutron and Muon Source, UK

Merlin neutron spectrometer at ISIS

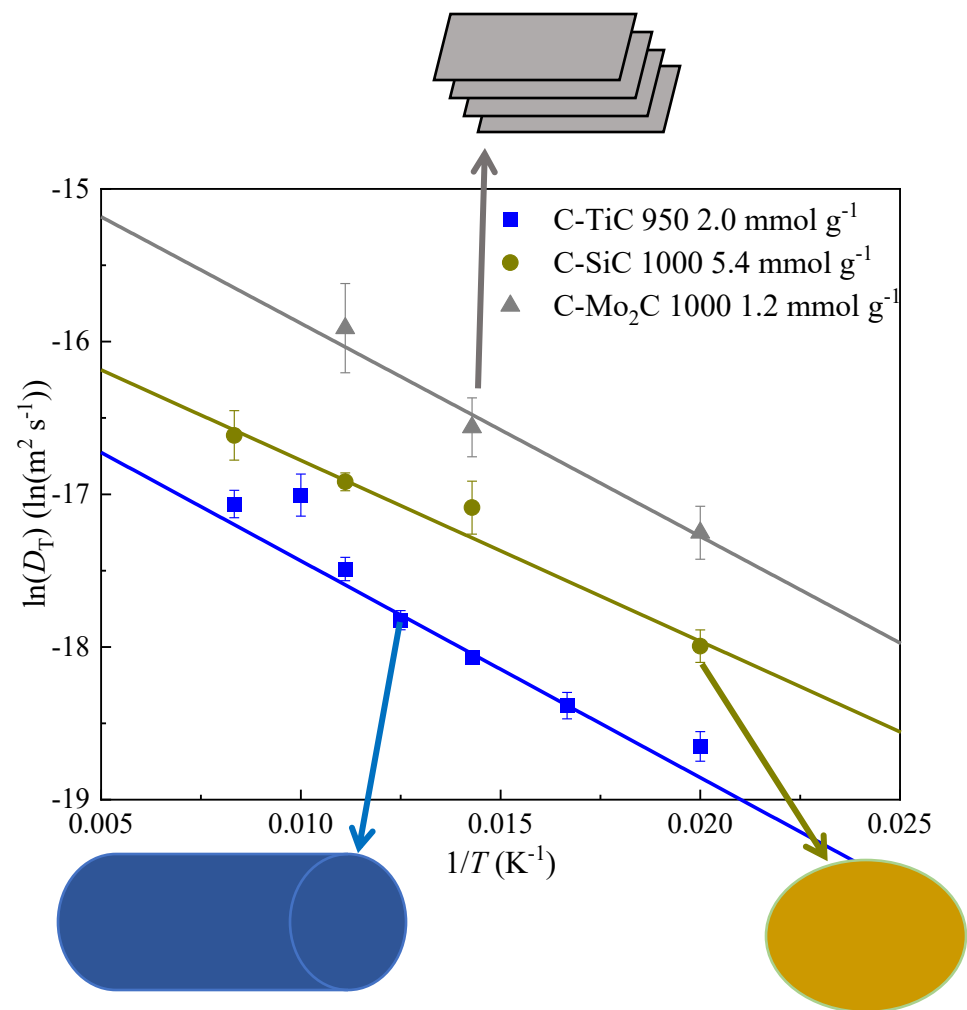
- Development of synthesis processes:
 - Use of sustainable carbon sources
 - Optimization of synthesis routines for optimal porosity

Combined modelling of three gas adsorption isotherms

M. Koppel, et al., Carbon 219, **2024**, 118799

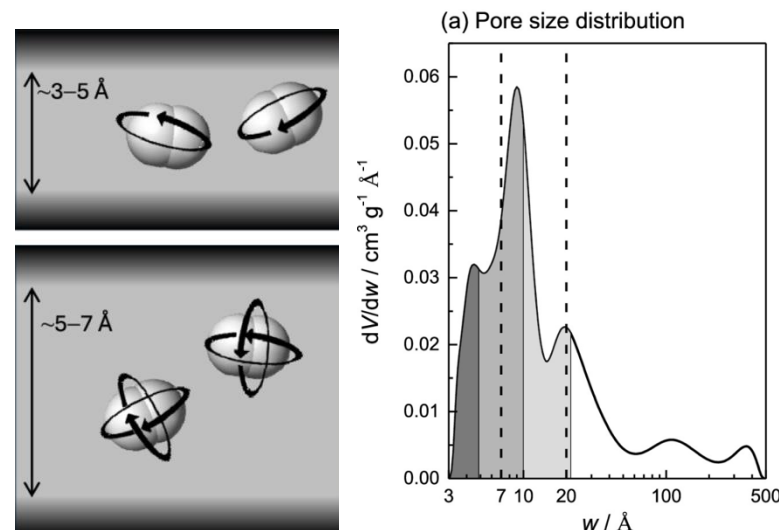
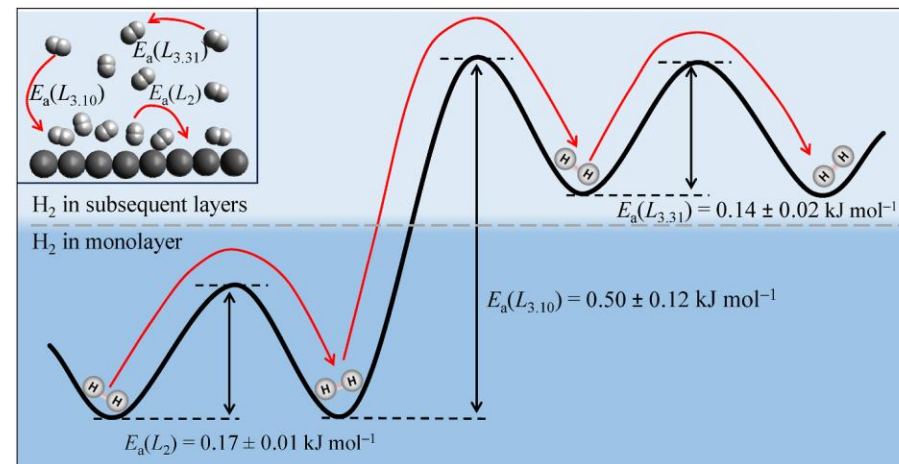
Different pores effect on H₂ confinement

Effect of pore geometry on H₂ confinement



R. Härmäs, et al., Carbon 155, **2019**, 122

H₂ mobility in between adsorption layers

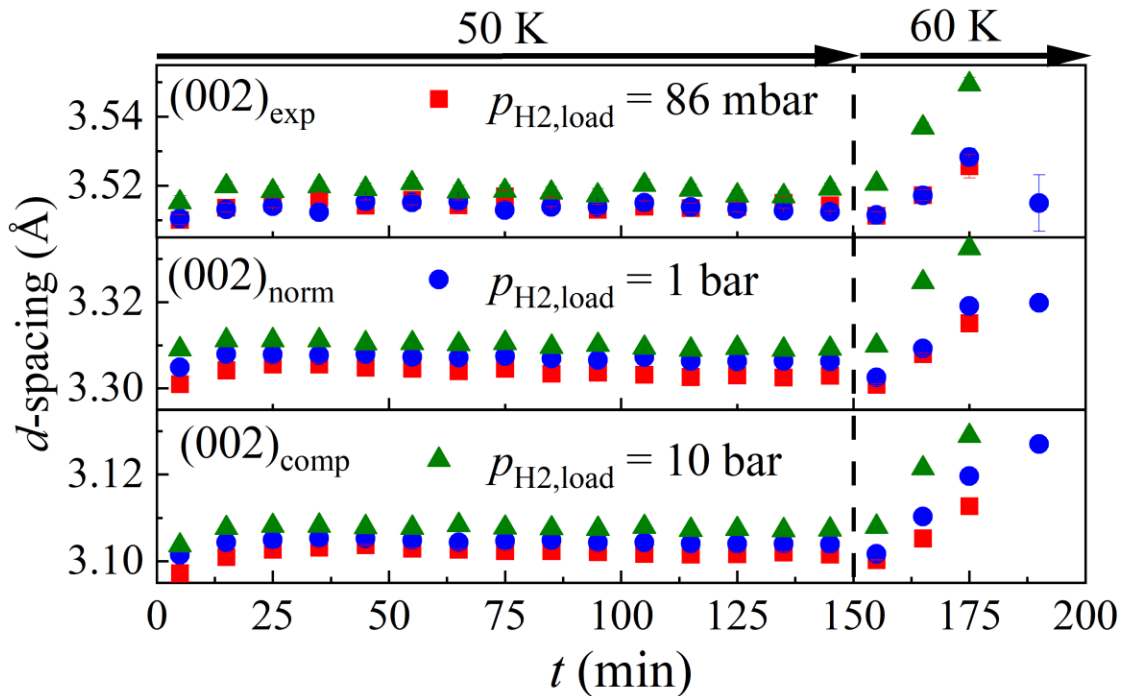
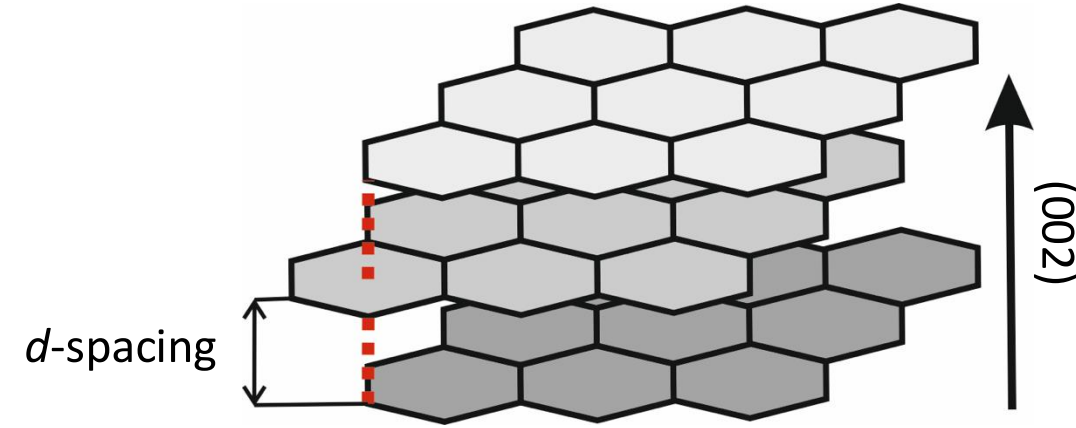


M. Koppel, et al., Carbon 219, **2024**, 118799

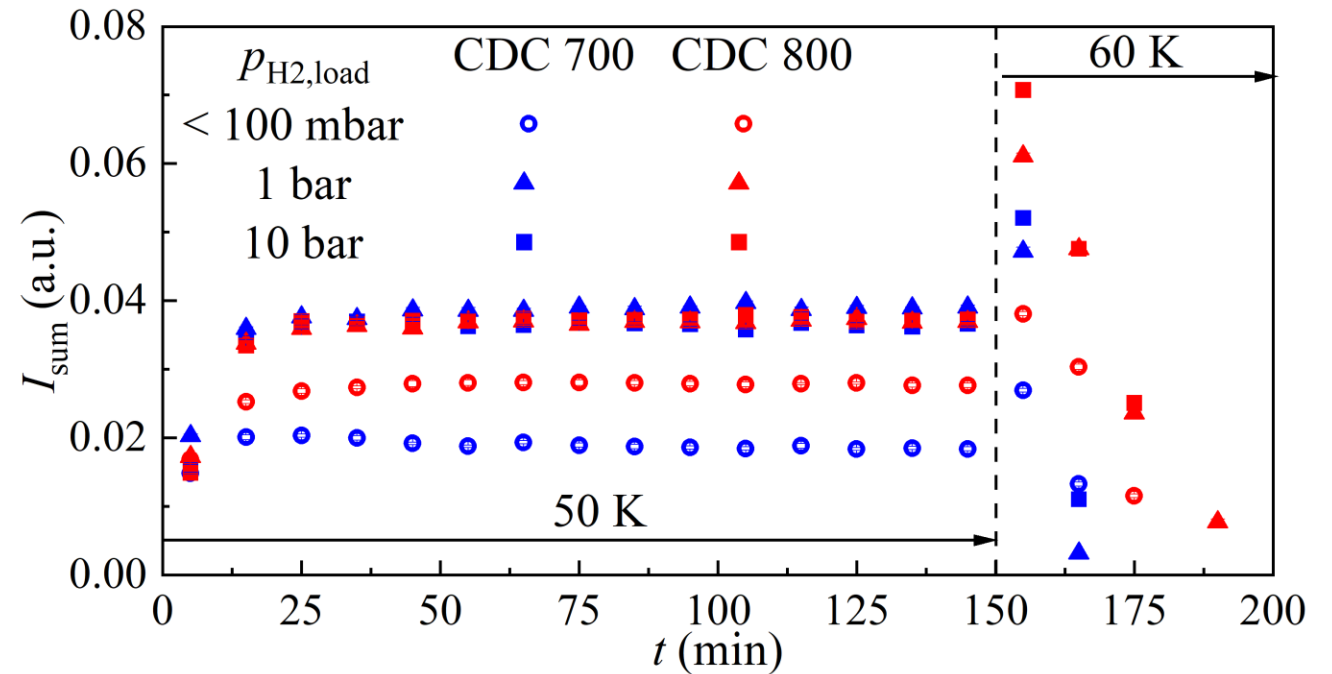
M. Koppel, et al., J. Phys. Chem. C 129, **2025**, 4789

Effect of strong H₂ confinement on adsorbent

- The graphenic carbon structure orders:
 - At 50 K and under H₂ loading
 - Original graphenic structure restored at 60 K
 - Formed structure possible through internal H₂ pressure



d -spacing of the three Bragg peaks Under H₂ loading

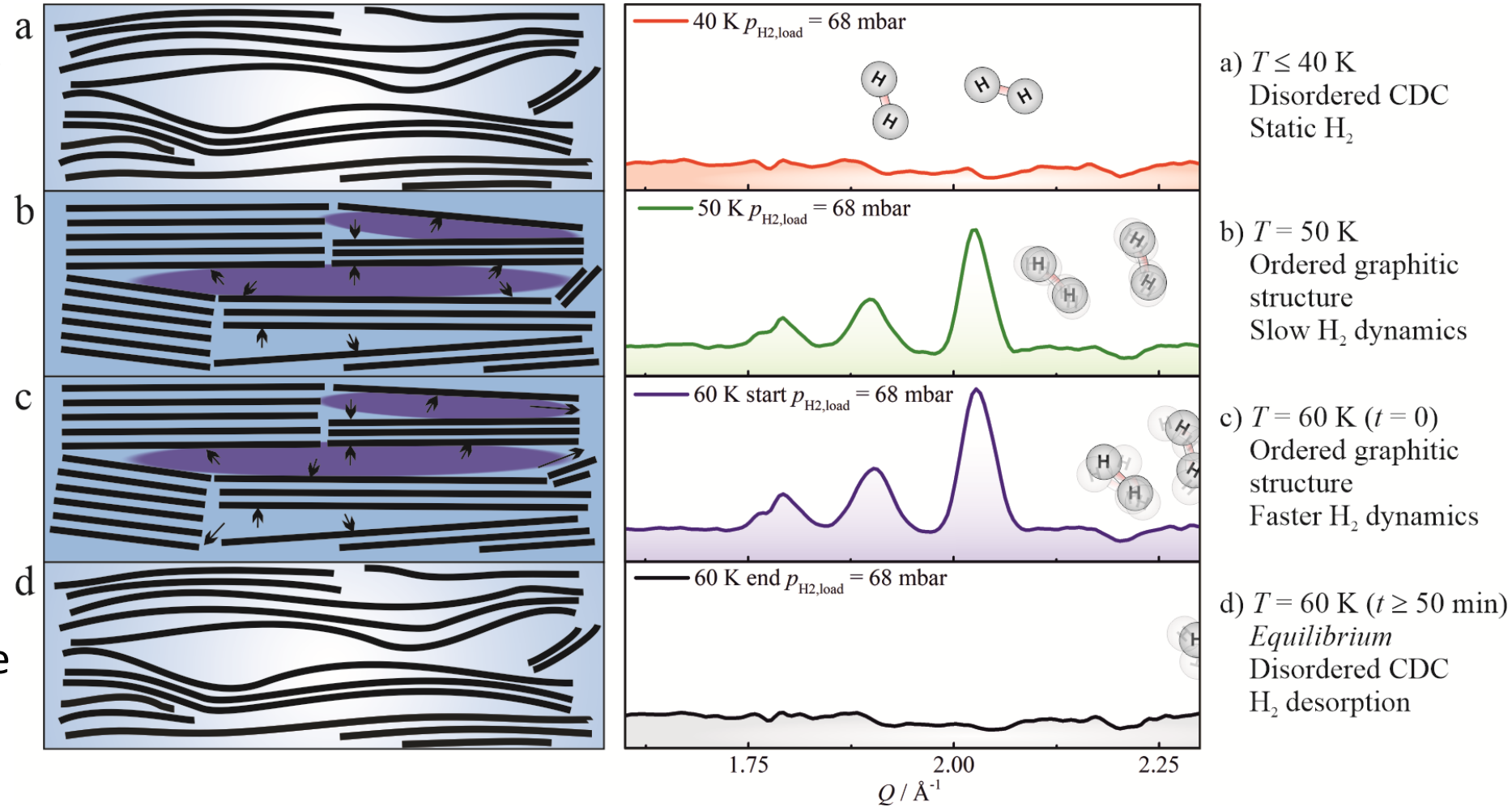


Summed up integrated intensities of the three Bragg peaks

Effect of strong H₂ confinement on adsorbent

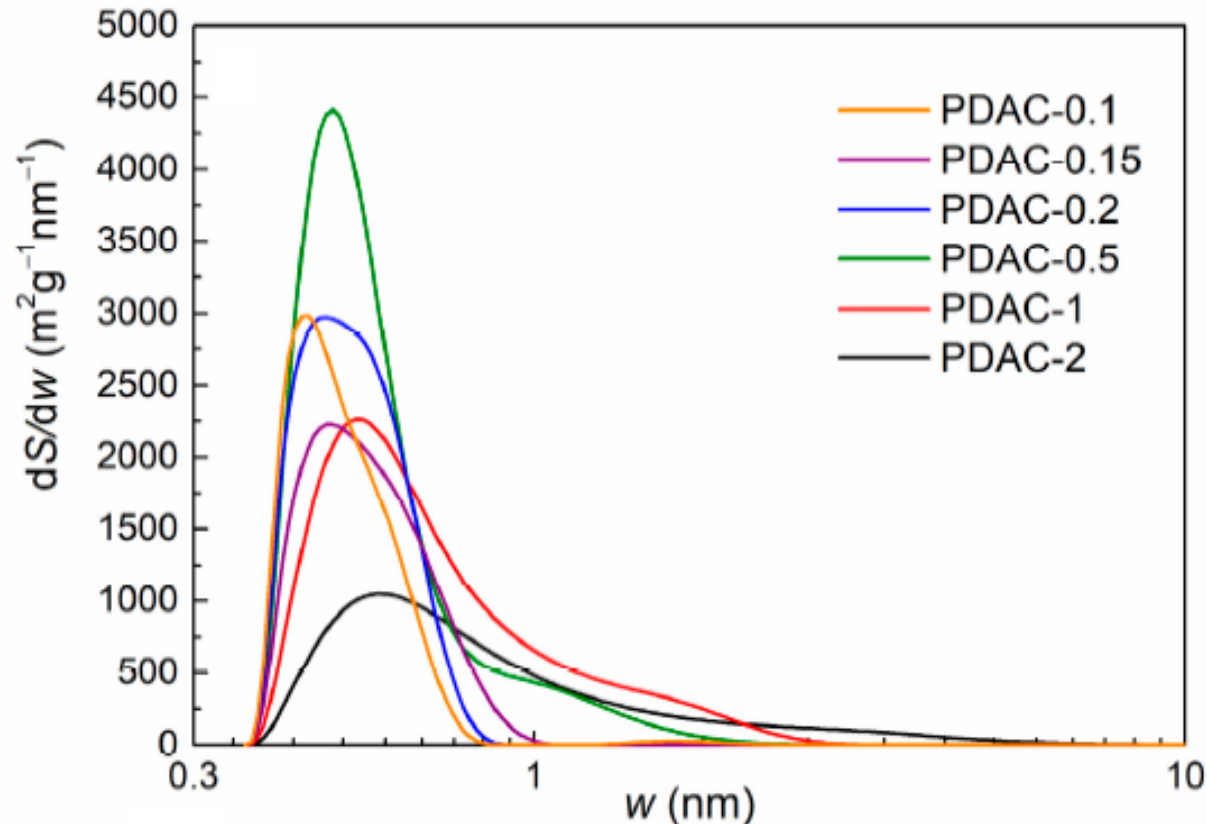
Reversible structural changes of adsorbent under H₂ loading:

- High density adsorbed H₂ phase at 20 K
- Kinetically limited desorption upon heating
- H₂ confined in ultramicropores causes internal pressure
- Upon further heating, additional internal pressure causes H₂ release and restoration of original structure

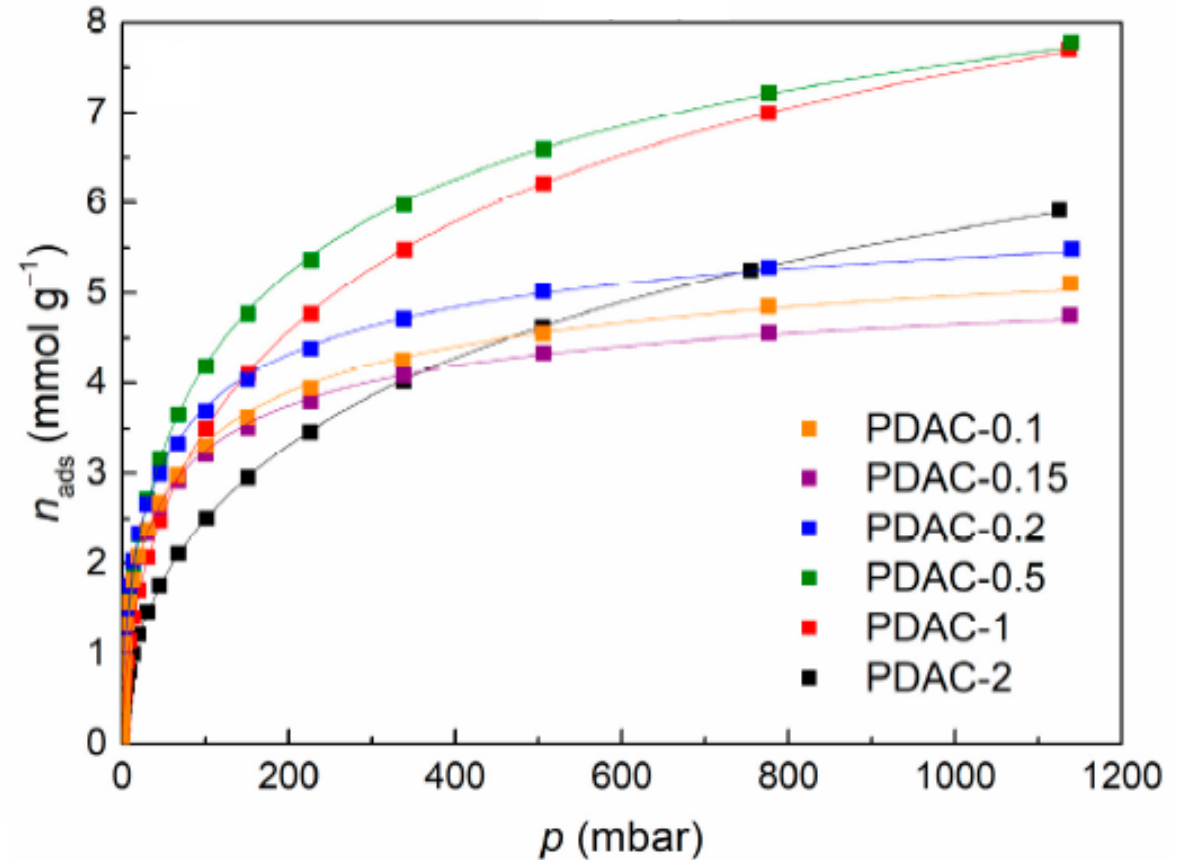


Synthesis of carbon-based H₂ adsorbents

- Dominantly ultramicroporous carbons
- Ultramicroporous carbons achieve surface fillment at lower pressure



Pore size distributions by specific surface area



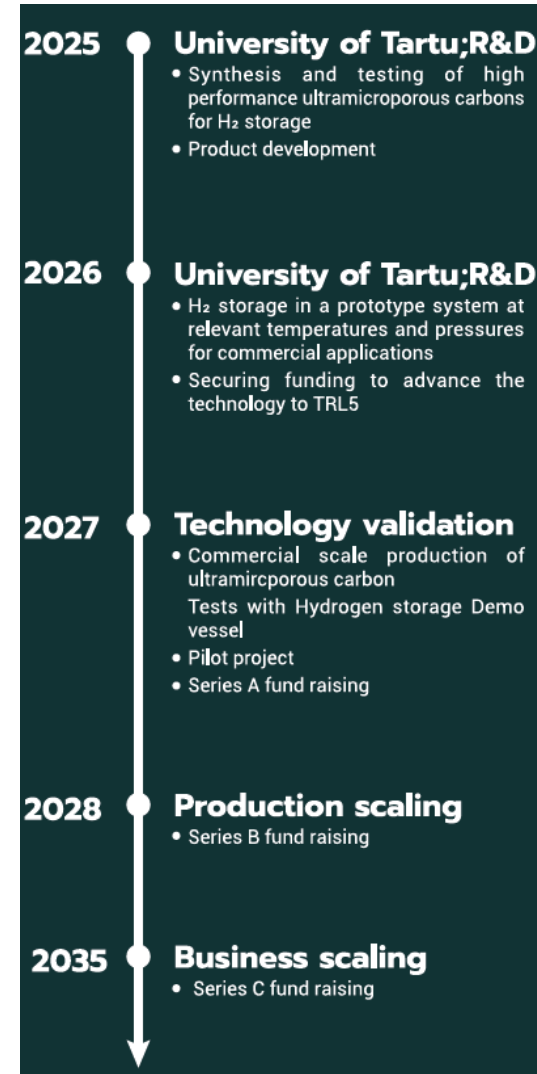
H₂ adsorption isotherms obtained at 77 K

Making ideas into reality



- Collaboration between Estanc AS and University of Tartu towards developing cryoadsorption based H₂ storage solutions
- Sub-enterprise H2Adsorb formed under Estanc AS for commercialization
- Goal of achieving H₂ storage density equivalent to 350 bar pressurization at:
 - 40 bar and -80 °C
- Lowered pressurization and tank requirements thank to ultramicroporous adsorbent
- Sustainable carbon-based adsorbents

<https://h2adsorb.eu/>



Prototype H₂ storage system by Renee Kauler, Christen Haamer, and Sander Kahk

Thank you for listening!

Would not be possible without wonderful collaborations



UNIVERSITY OF TARTU
Institute of Chemistry

- Egert Möller
- Miriam Koppel
- Riinu Härmas
- Kenneth Tuul
- Maarja Paalo
- Tavo Romann
- Thomas Thomberg
- Heisi Kurig
- Enn Lust
- Many more



- Martin Månsson
- Frank Elson



- Jacek Jagiello



UNIVERSITY OF TARTU
Department of Geology

- Jaan Aruväli



Science and
Technology
Facilities Council

ISIS Neutron and
Muon Source

- Mark Telling
- Manh Tu Le
- Tatiana Guidi



- Eneli Monerjan
- Margarita Russina
- Veronika Grzimek

Funding acknowledgment from projects: TK141; TK210; PUTJD957; PRG676; PSG935, TEM-TA81

